

Poluanții. Procedee de degradarea a acestora. Toxicitate.

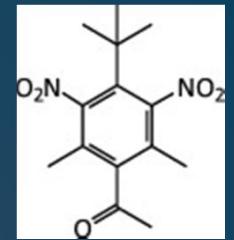
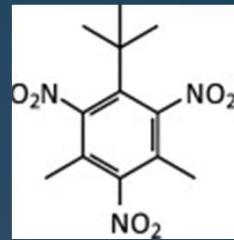
**Rezultate profesionale anterioare și
planuri pe termen scurt**

Dr. Mariana Neamțu

Metode de oxidare avansate (AOPs)

Poluanți/ Micropoluanți	Specii reactive ale oxigenului (ROS) rezultate din precedeele:	Producții de degradare	Teste de toxicitate
Nitromoscuri (nitromosc xilen (NMX), nitromosc cetonă (NMK), nitromosc ambret (NMA), nitromosc mosken (NMM), nitromosc tibeten (NMT))	UV UV/H_2O_2 O_3 UV/O_3 $Fe(II)/H_2O_2$ $UV/Fe(II)/H_2O_2$ $solar/Fe(II)/H_2O_2$ $UV/Fe(II)/H_2O_2$	hidrochinona 2-hidroxi-4-(Nacetyl) Aminofenol acetamida p-benzochinona fenolul acidul oxamic acidul glicolic acidul tartronic acidul maleic acidul glioxilic acidul ketomalonic acidul oxalic acidul acetic acidul formic etc	Testul de bioluminescență (<i>Vibrio fisheri</i>) Testul YES (Yeast Estrogen Screen)
Bisfenolul A (BPA)			
Coloranți textili (RY84, RR120, RB5, Disperse Red 354, Procion Marine H-EXL)	$UV/Cu(II)/Fe(III)/H_2O_2$ $UV/Fe(III)oxalat/H_2O_2$ $UV/Fe(III)/H_2O_2$ $dark/Fe(II)/H_2O_2$ $solar/Fe(III)oxalat/H_2O_2$ $UV/Fe(II)$		
Alchilfenoletoxilați Nonilfenol Octilfenol	$Zeoliti$ ($FeY5$, $FeY11.5$, $FeY80$) $Al-Fe PILC$		
Medicamente (atenolol, claritromicina, gabapentin, metformin, metoptolol, primidon, paracetamol)			
Alți compuși chimici (benzotriazol, polifenoli, metilbenzotriazol etc)			

Nitromosuri



Nitomosc xilen

Nitomosc cetonă



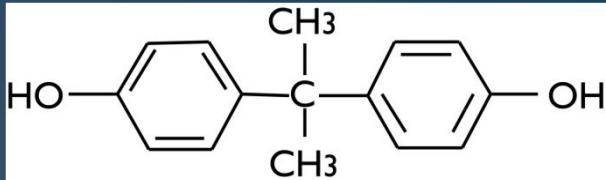
Synthetic Musks

Neamtu, M., Siminiceanu I., Kettrup A., (2000) *Chemosphere*, 40(12), 1407-1410.

Neamtu M., Sminiceanu I., (1999) *Rev. Chim. (Bukarest)*, 50(7) 545-553.

Neamtu M., Sminiceanu I., (1998) *Rev. Chim. (Bukarest)*, 49(11), 745-750.

Bisfenol A



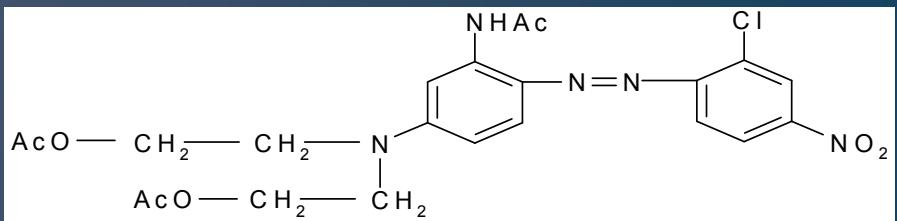
Tercero Espinoza, L.A., Neamtu M., Frimmel F. H. (2007), *Water Research*, 41, 4479-4487
Neamtu M., Frimmel F. H. (2006), *Water Research*, 40, 3745-3750
Siminiceanu I., Neamtu M., (2001), *Rev. Chim. (Bucureşti) (English Edition)*, 2 (1-2) 19-22

Coloranți textili

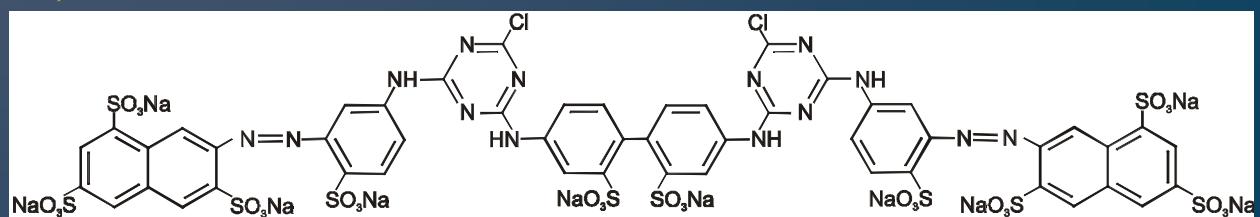
- Neamtu M., Catrinescu C., Kettrup A. (2004), *Applied Catalysis: Environmental*, 51, 149-157.
- Neamtu M., Yediler A., Siminiceanu I., Macoveanu M., Kettrup A. (2004), *Dyes and Pigments*, 60, 61-68.
- Neamtu M., Zaharia C., Catrinescu C., Yediler A., Macoveanu M., Kettrup A. (2004), *Applied Catalysis: Environmental* 48, 287-294.
- Neamtu M., Yediler A., Siminiceanu I., Kettrup A. (2003), *Journal of Photochemistry and Photobiology A: Chemistry*, 161 (1), 87-93.
- Neamtu, M., Siminiceanu I., Yediler A., Kettrup A., (2002), *Dyes and Pigments*, 53, 93.
- Catrinescu C., Neamtu M., Yediler A., Macoveanu M., Kettrup A. (2002), *Environmental Engineering and management Journal*, 1(2), 177.
- Siminiceanu I., Neamtu M., (2001), *Rev. Chim. (Bucureşti) (English Edition)*, 2 (1-2) 19
- Lienert D., Neamtu M., Koch M., Yediler A., Kettrup A., (2000), *Melliand Textilberichte International Textile Reports*, 81(6), 523-525
- Neamtu, M., Yediler A., Siminiceanu I., Macoveanu M., Kettrup A., (2000), *Toxicological and Environmental Chemistry*, vol.78, 31-40



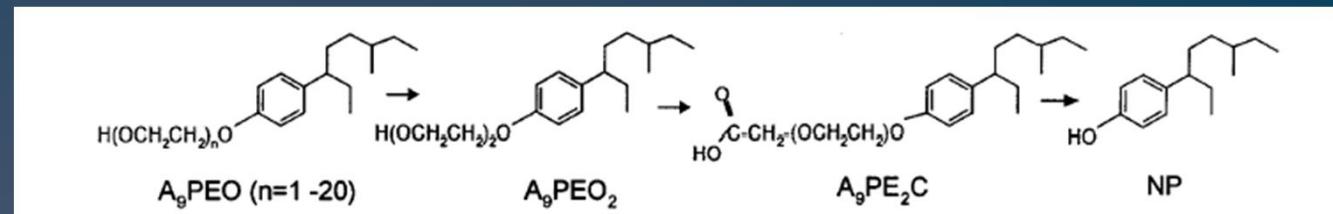
Disperse Red 354 (CI)



Reactive Yellow 84 (CI)



APEO



Neamtu M., Popa D. M., Frimmel F. H. (2009), *Journal of Hazardous Materials*, 164, 1561

Neamtu M., Popa D. M., Frimmel F. H. (2009). Photodegradation of octylphenol using simulated and natural sunlight radiation. In *Sewage treatment: uses, processes and impact*, Eds. A. Stephens and M. Füller, Nova Science Publishers, New York, ISBN 978-1-60876-875-2, pp. 341-362.

Tercero Espinoza, L.A., Neamtu M., Frimmel F. H. (2007), *Water Research*, 41, 4479

Neamtu M., Frimmel F. H. (2006), *Water Research*, 40, 3745

Neamtu M., Frimmel, F. H. (2006), *Science of the Total Environment* 369 (1-3), 295

Jahnel J., Neamtu M., Schudoma D., Frimmel F. H. (2006), *Acta hydrochemica et hydrobiologica* 34 (4), 389

Neamtu M., Popa D. M., Frimmel F. H. (2006) Photodegradation of endocrine disrupting chemicals nonylphenol and octylphenol in River Rhine and Lake Hohloh waters. In *Humic Substances – Linking Structure to Functions*. Eds. Fritz H. Frimmel, Gudrun Abbt-Braun, Universität Karlsruhe, ISSN 1612-118X, 45-II, p.913-917

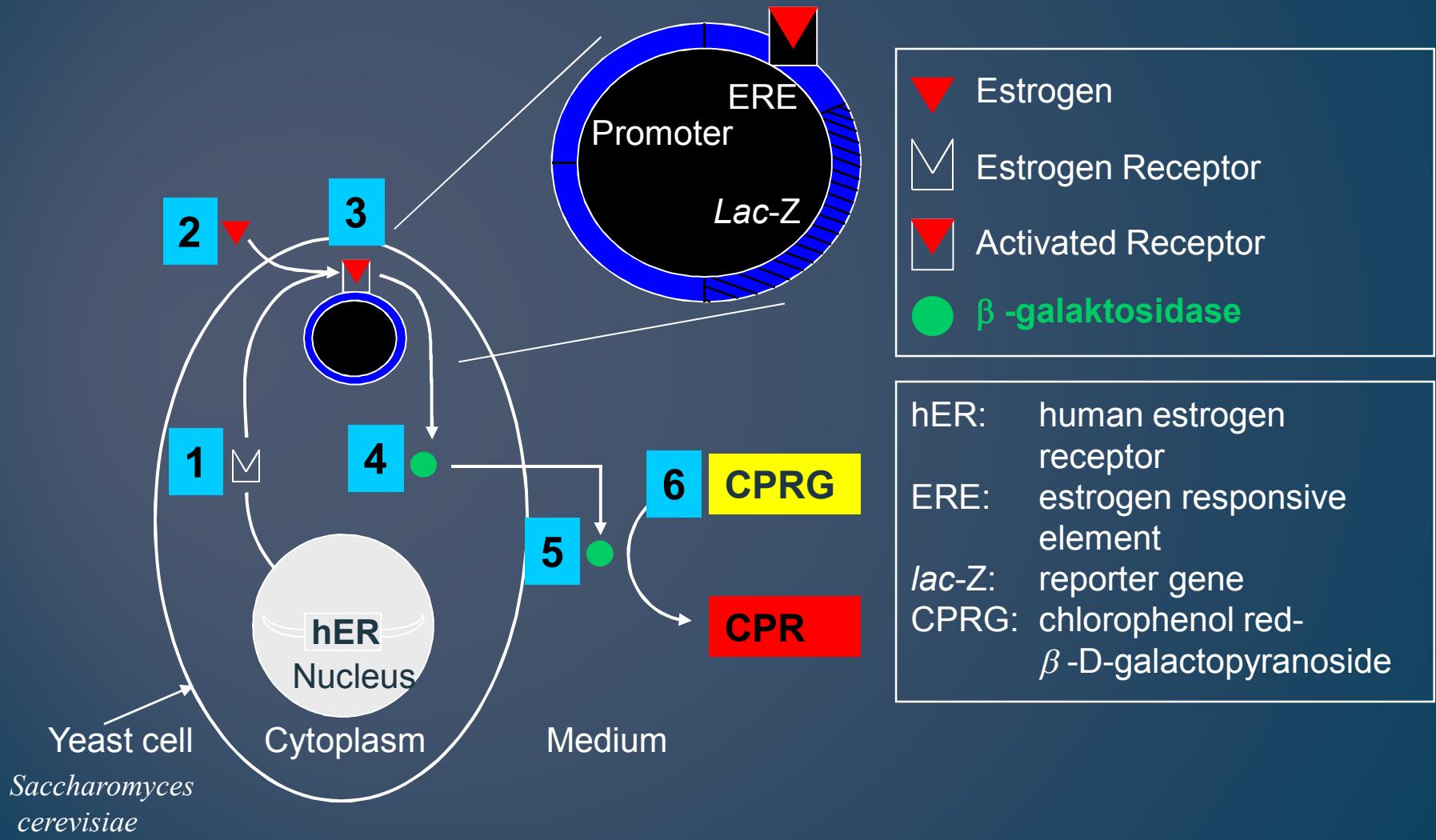
Tehnici de laborator utilizate



Metode cromatografice (HPLC, HPICE,
TLC, GC/MS)
UV/VIS , ICP-AES, TOC,
XRD, FTIR, XPS
Teste de ecotoxicitate și cancerigenitate

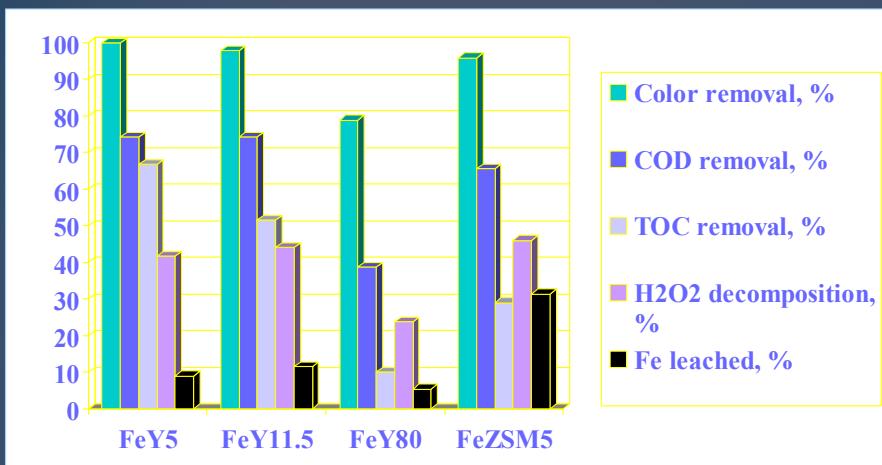


Testul de cancerigenitate (YES)



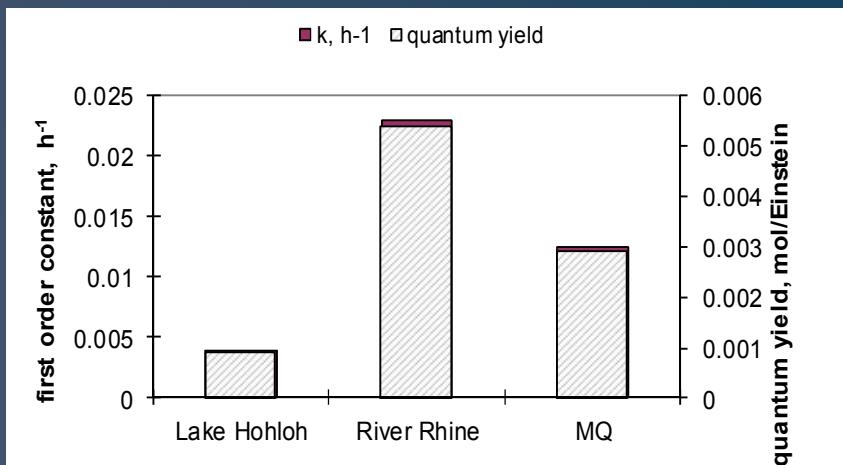
Routledge, E. J., Sumpter, J. P. (1996), Environ. Toxicol. Chem. 15, 241-248

Câteva rezultate



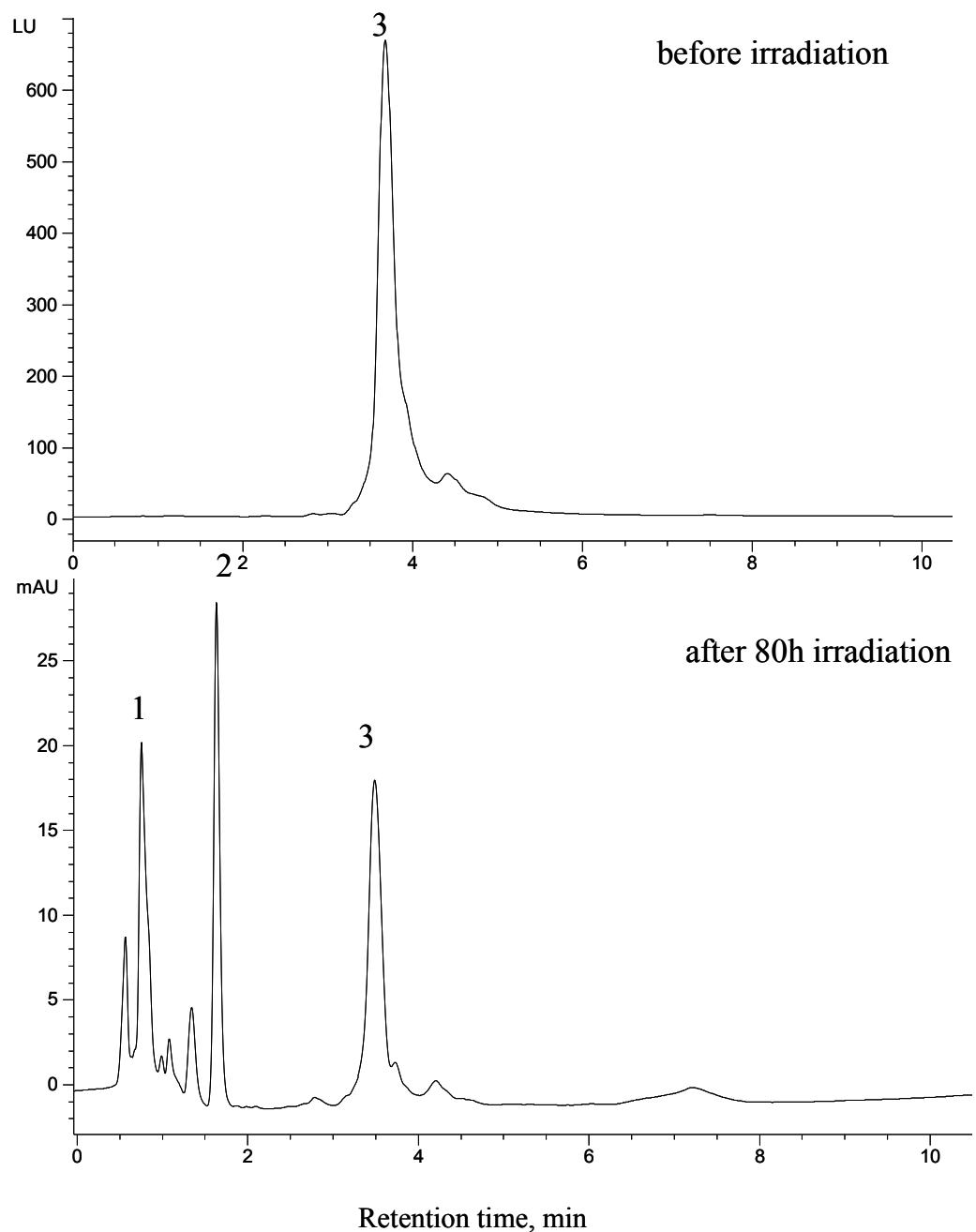
Catalytical removal of RY84. Initial conditions: $C^0(\text{dye}) = 100 \text{ mg/L}$, $\text{pH}=5$, $C^0(\text{H}_2\text{O}_2) = 20\text{mM}$, $C^0(\text{catal})=1\text{g/L}$ and reaction time of 120 min.

Neamtu M., Catrinescu C., Kettrup A. (2004), *Applied Catalysis: Environmental*, 51, 149-157.
Neamtu M., Zaharia C., Catrinescu C., Yediler A., Macoveanu M., Kettrup A. (2004), *Applied Catalysis: Environmental* 48, 287-294



Effect of DNOM upon photodegradation of nonylphenol. Initial conditions: $2.55 \times 10^{-5} \text{ M NP}$, 10°C

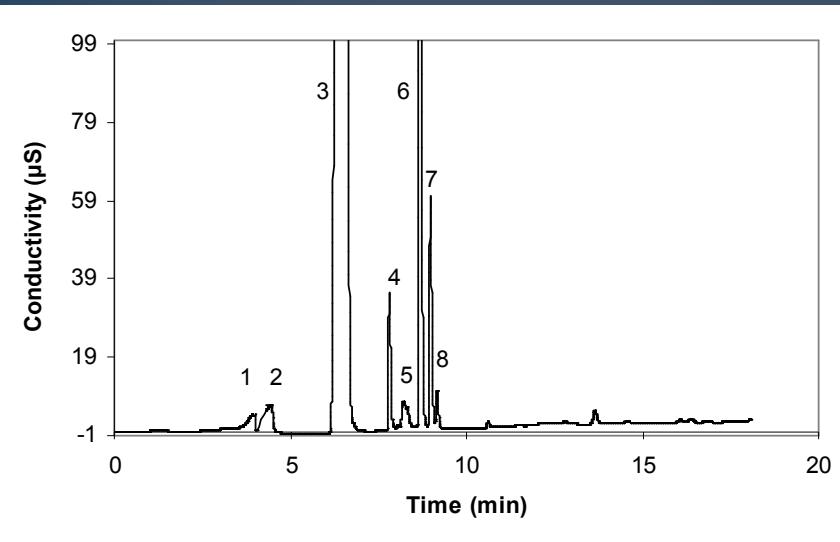
Neamtu M., Frimmel, F. H. (2006), *Science of the Total Environment* 369 (1-3), 295-306



HPLC chromatogram of nonylphenol and its intermediates (phenol and 1,4-dihydroxylbenzene).
Initial conditions: $2.55 \times 10^{-5} M$ of NP, pH 5.4, 10 °C, 50 mM H_2O_2 . Peak identities are as follows: (1) DHB; (2) phenol; (3) nonylphenol

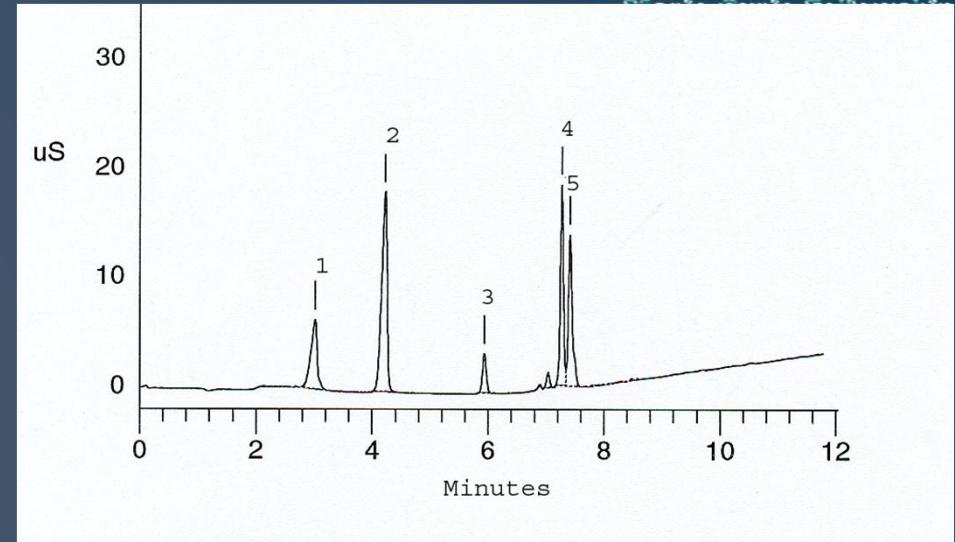
Neamtu M., Frimmel, F. H.
(2006) Photodegradation of
endocrine disrupting chemical
nonylphenol by simulated solar
UV-irradiation, *Science of the
Total Environment* 369 (1-3),
295-306

Producții degradării HPICE



Procion Marine H-EXL after 60 minutes of the catalytic oxidation.

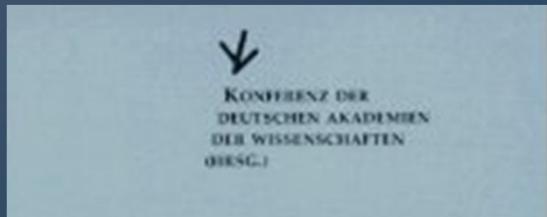
Applied Catalysis: Environmental 48, (2004), 287.



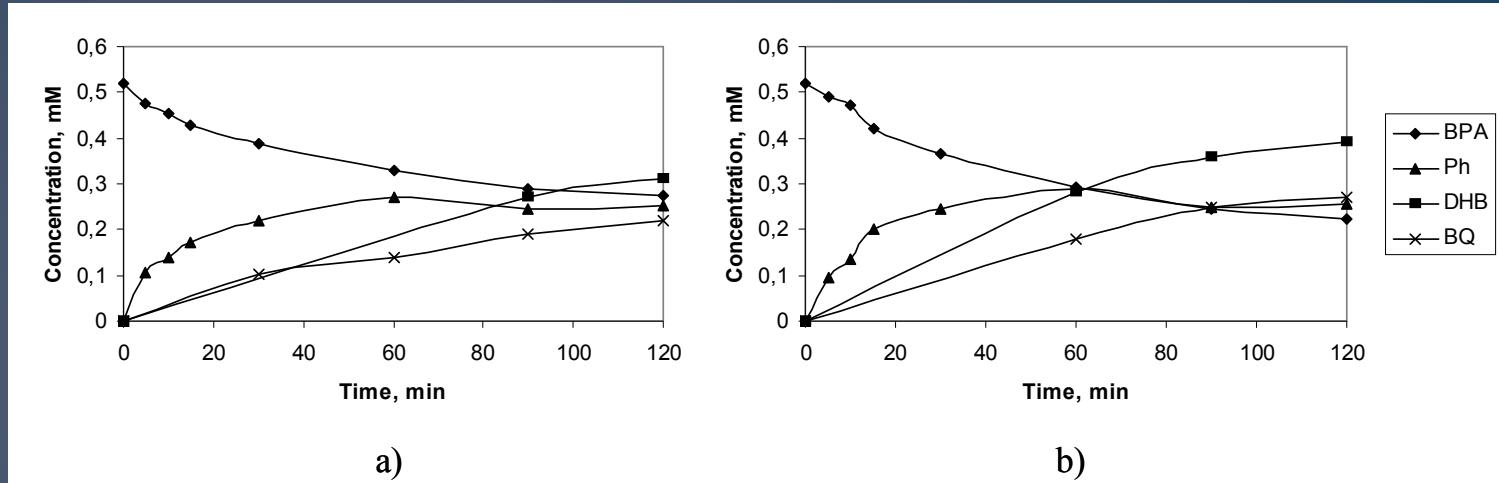
*Disperse Red 354 after 30 minutes irradiation with 24.5 mmol $\text{H}_2\text{O}_2/\text{L}$.
Peak identities are as follows: 1, formate; 2, chloride; 3, nitrate; 4, sulphate; 5, oxalate.*

Dyes and Pigments, 60, (2004), 61.

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Producții degradării



Degradation of Bisphenol A in MilliQ water and formation of products upon irradiation at 253.7 nm in the presence of 250 $\mu M H_2O_2$ (a) and 750 $\mu M H_2O_2$ (b)

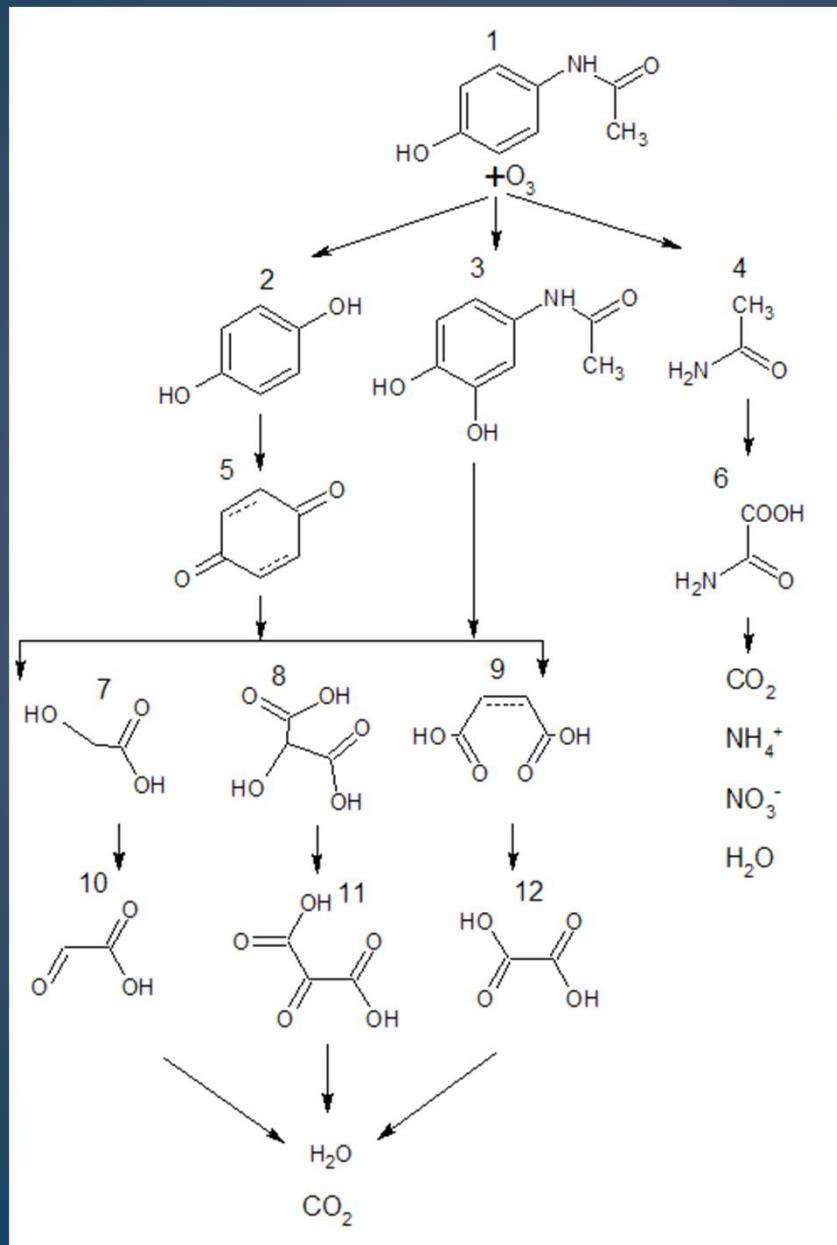
Neamtu M., Frimmel F. H. (2006) Degradation of endocrine disrupting Bisphenol A by 254 nm irradiation in different water matrices and effect on yeast cells, *Water Research*, 40, 3745-3750



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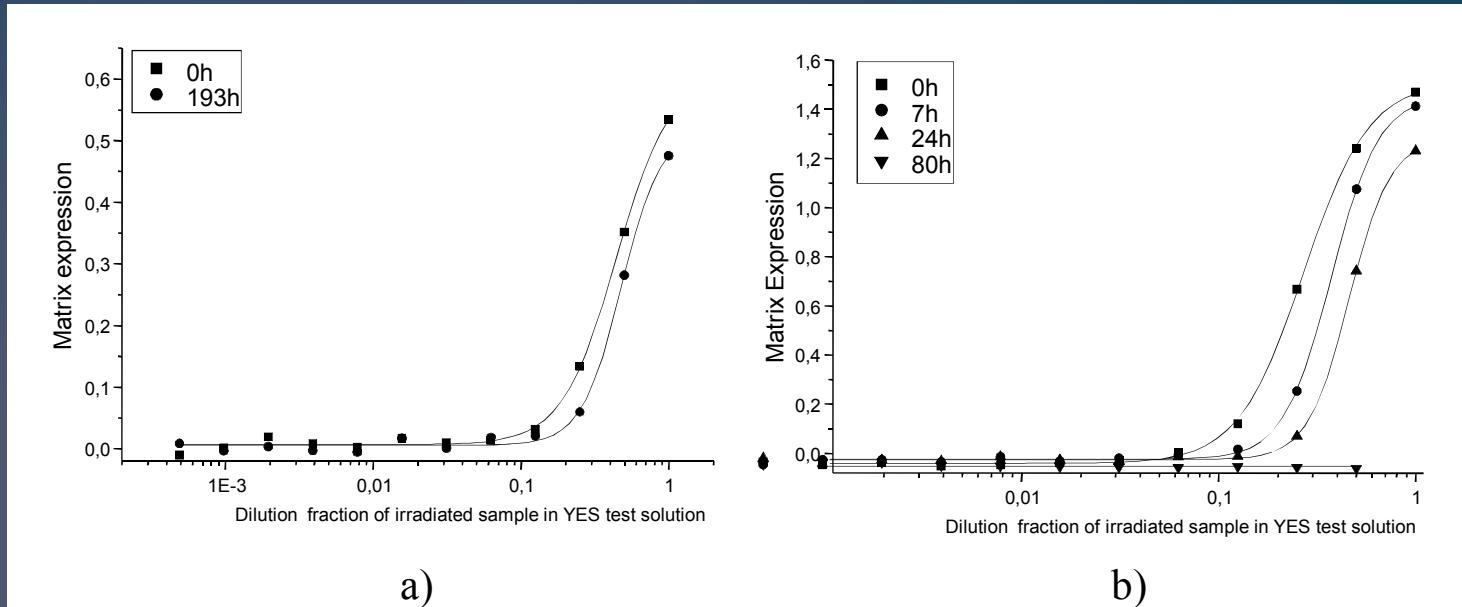
Mecanismul degradării



The possible mechanism of the degradation of paracetamol (1—paracetamol, 2—hydroquinone, 3—2-hydroxy-4-(Nacetyl) aminophenol, 4—acetamide, 5—p-benzoquinone, 6—oxamic acid, 7—glycolic acid, 8—tartronic acid, 9—maleic acid, 10—glyoxylic acid, 11—ketomalonic acid, 12—oxalic acid).

Neamtu M., Bobu M., Kettrup A., Siminiceanu I. (2013). *Journal of Environmental Science and Health, Part A*, 48 (10), 1264 – 1271.

Testul YES



Yeast screening of estrogen activity for experiments under solar simulator irradiation in the absence (a) and in presence of 50 mmol/l H_2O_2 (b). Initial conditions: $2.55 \times 10^{-5} \text{ M}$ of nonylphenol, pH 5.4, 10°C. The axis X represents the dilution fraction of tested sample across 12 wells in a 96-well plate (undiluted – 1, second – 0.5, third – 0.25 etc)

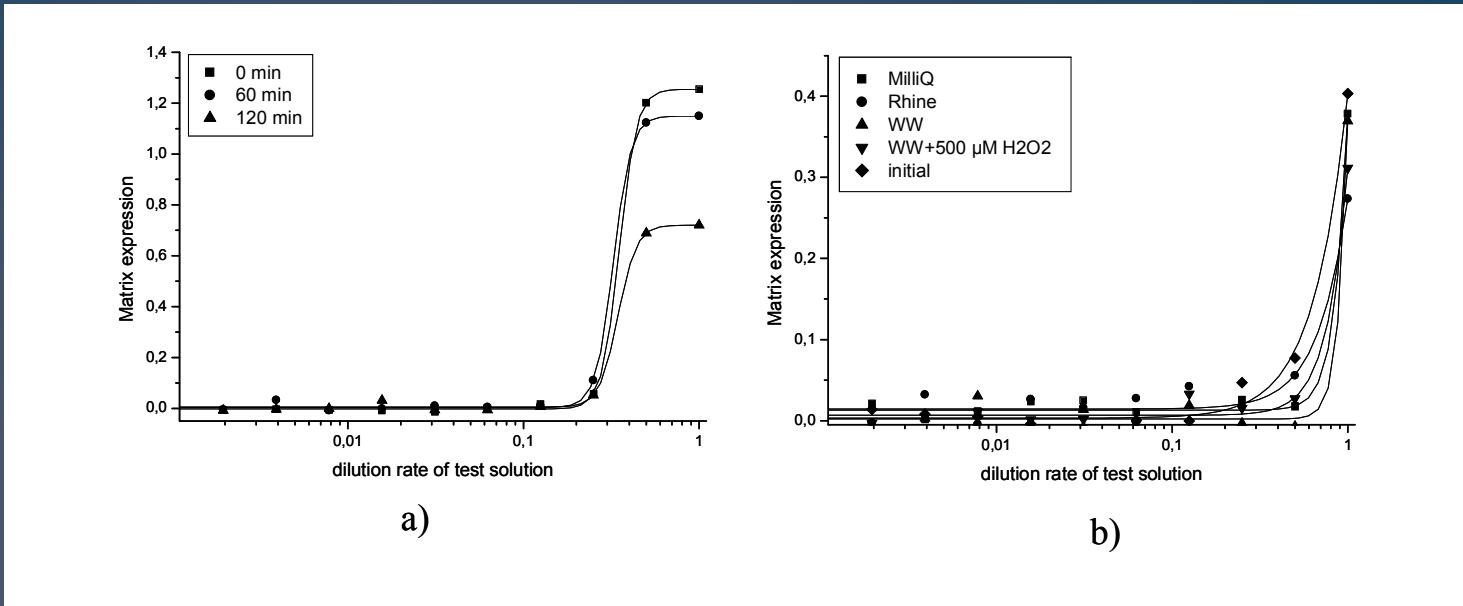
Neamtu M., Frimmel, F. H. (2006) Photodegradation of endocrine disrupting chemical nonylphenol by simulated solar UV-irradiation, *Science of the Total Environment* 369 (1-3), 295-306



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Testul YES



Response of estrogen screen for irradiated samples. (a) Disappearance of estrogenic activity of bisphenol A in MilliQ water with irradiation time; (b) Disappearance of estrogenic activity of bisphenol A in different waters, irradiation time 120 min. Initial conditions: 520 μ M of bisphenol A, 500 μ M H₂O₂. The X axis represents the dilution fraction of tested sample across 12 wells in a 96-well plate (undiluted – 1, second – 0.5, third – 0.25, etc.)

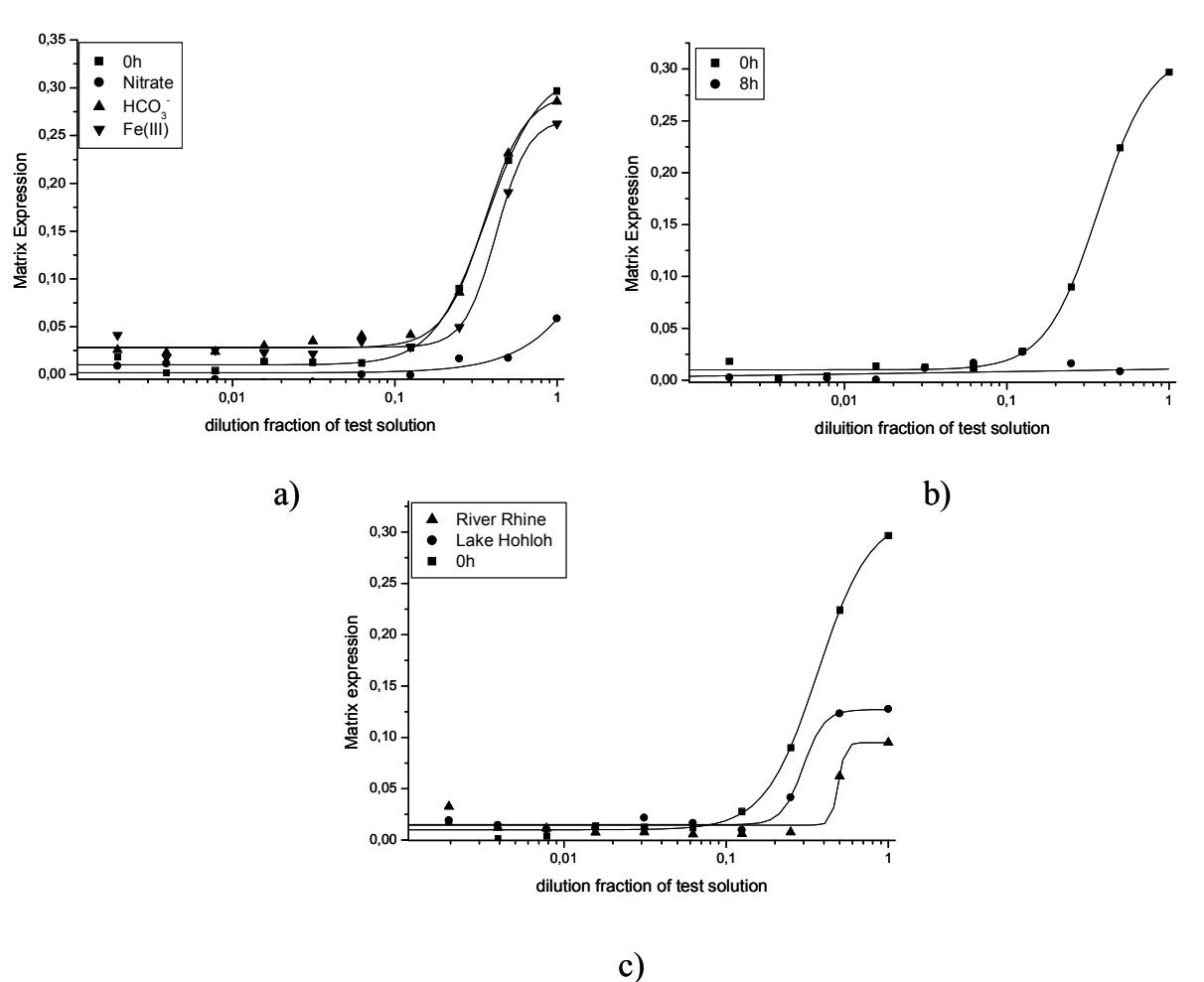
Neamtu M., Frimmel F. H. (2006) Degradation of endocrine disrupting Bisphenol A by 254 nm irradiation in different water matrices and effect on yeast cells, *Water Research*, 40, 3745-3750



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Testul YES



Yeast screening of estrogen activity of OP under solar simulator irradiation for experiments: (a) – under solar irradiation in the initial presence of 61 mg NO₃⁻/L, 100 µg Fe (III)/L, 750 mg HCO₃⁻/L (b) – under solar simulated irradiation in the presence of 50 mM H₂O₂, (c) – under solar radiation, in the presence of Rhine River and Hohloh Lake water. Irradiation time 8h. The axis X represents the dilution fraction of tested sample across 12 wells in a 96-well plate (undiluted – 1, second – 0.5, third – 0.25 etc)

Neamtu M., Popa D. M., Frimmel F. H. (2009).
Journal of Hazardous Materials, 164, 1561-1567

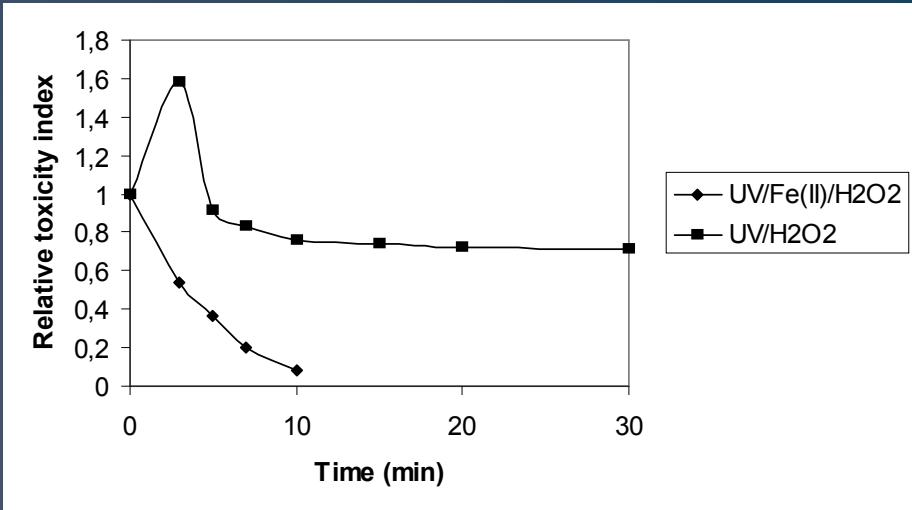
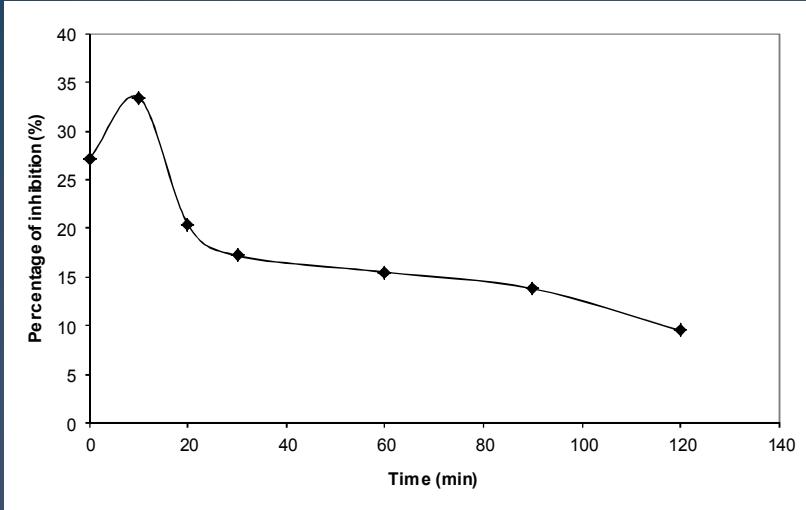


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Testul de bioluminescență



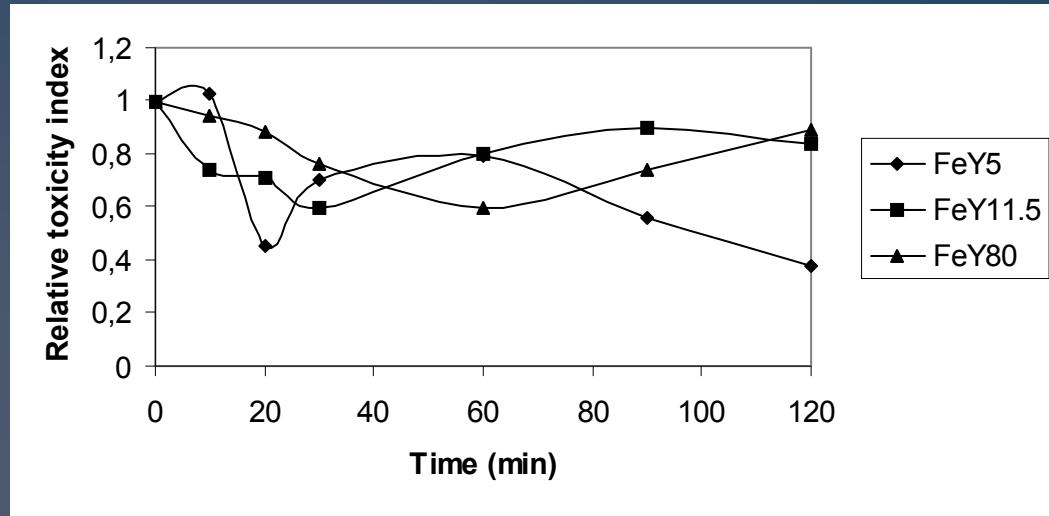
Percentage inhibition after 15 minutes incubation by the bacterium *Vibrio fisheri* in LUMISTox300 of *Procion Marine H-EXL* azo-dye. Initial conditions were 100 mg/L azo-dye, pH=5, t=50°C, catalyst concentration 1 g/L and 20 mmol/L H₂O₂.

Neamtu M., Zaharia C., Catrinescu C., Yediler A., Macoveanu M., Kettrup A. (2004), Applied Catalysis: Environmental 48, 287-294.

Neamtu M., Yediler A., Siminiceanu I., Kettrup A. (2003), Journal of Photochemistry and Photobiology A: Chemistry, 161 (1), 87-93.

Toxicity assay using the bacterium *Vibrio fisheri* in LUMISTox 300 during the treatment. Incubation time 30 minutes. Initial conditions: C⁰(dye) = 100 mg/L, C⁰(H₂O₂) = 5 mM, C⁰(Fe²⁺) = 0.25 mM

Testul de bioluminescență



Toxicity assay using the bacterium *Vibrio fisheri* in LUMISTOX 300 during the treatment with different catalysts. Incubation time 30 minutes. Initial conditions were 100 mg/L azo-dye, pH=5, t=50°C, catalyst concentration 1 g/L and 20 mmol/L H₂O₂.

Neamtu M., Catrinescu C., Kettrup A. (2004) Effect of dealumination of iron (III) - exchanged Y zeolites on oxidation of reactive yellow 84 azo dye in the presence of hydrogen peroxide, *Applied Catalysis: Environmental*, 51, 149-157.



Alte teme/proiecte cercetate

- German Countries Working Party on Water (Länderarbeitsgemeinschaft Wasser – LAWA) LAWA-Nr. 0 10.03; 2003-2006. Tema: „Recommendation for Environmental Quality Standards for the protection of aquatic biota in surface waters”; *Membru în echipă*.

Jahnel J., Neamtu M., Schudoma D., Frimmel F. H. (2006) Bestimmung von Umweltqualitätsnormen für potenziell gewässerrelevante Stoffe. *Acta hydrochemica et hydrobiologica* 34 (4), 389-397

- Grant Romano-Elvețian, finanțat de Swiss National Science Foundation (SNSF), ESTROM, IB6120-106996-POPIASI. 2005-2007. Tema: “Chemical, Biological and Ecotoxicological Assessment of the Bahlui River, in North-Eastern Romania. Case Study: Pesticides and Persistent Organic Pollutants (POPs)”; *Co-Director de proiect*.

Ciumasu M.I., Costica M., Costica N., Neamtu M., Dirtu A. C., De Alencastro L. F., Buzdugan L., Andriesa R., Iconomu L., Stratu A., Popovici O. A., Secu C. V., Olariu C. P., Dunca S., Stefan M., Lupu A., Stingaciu-Basu A., Netedu A., Dimitriu R.-I., Gavrilovici O., Talmaciu M., Borza M. (2012)., *Journal of Hazardous, Toxic, and Radioactive Waste* (ASCE), 16 (2), 158-170.

Neamtu M., Ciumasu I. M., Costica N., Costica M., Bobu M., Nicoara M. N., Catrinescu C., Becker van Slooten K., De Alencastro L. F. (2009), *Environmental Science and Pollution Research*, 16, S76

Ciumasu M.I, Neamtu M., Costica N., De Alencastro F. (2008), *Danube News*, 17, 9-10.

Ciumașu I. M., Costică N., Neamțu M., De Alencastro F. (2008), *Geo-Eco-Marina*, 14, 57-61.

- Grant finanțat de Swiss National Science Foundation (SNSF), IZKOZ2_L45025. 2013. “Degradation of 5 emergent contaminants on Fe photoresponsive catalysts”. *Director de proiect*.
- Ministerul Educației și Cercetării, Grant PN-II-ID-PCE-2011-3-0652. 2011-2014. “Granița romană în estul Provinciei Dacia”. *Membru în echipă*

Planuri pe termen scurt

PN-II-ID-PCE-2012-4-0477

